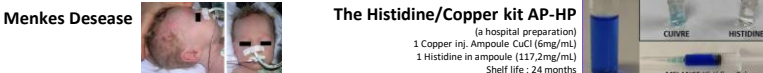


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## INTRODUCTION

**DRDP-AGEPS** → Développement de médicaments pour des maladies orphelines non couvertes par l'industrie concurrentielle (ex: Le kit Histidine/Cuivre pour la maladie de Menkes)



## OBJECTIVES

- Demonstrate the stability indicator nature of the degradation products (DP) research method of histidine ampoules.
- Identify the pathways of Histidine degradation and generate DP of the active substance (Histidine).
- Demonstrate the benefit of inerting in the production of Histidine ampoules: selection of a PDD representative of the degradation, the evolution of which is evaluated with a predictive method of stability.

## METHODS

### Forced degradation under the following conditions:

Stress	Experimental conditions	Storage conditions	Sampling time (days)	
Thermal	Amber glass vials Heat chamber	90°C	1, 2	
Photolytic	Transparent glass vials UV-Vis	25°C	8, 13, 20	
Oxidation	Amber glass vials Heat chamber	0,1 to 3% H <sub>2</sub> O <sub>2</sub> 1 to 5% MMPP	60°C 60°C	1 1
Hydrolysis	Amber glass vials Heat chamber	1M NaOH 1M HCl	60°C 60°C	8, 29, 57 8, 29, 57

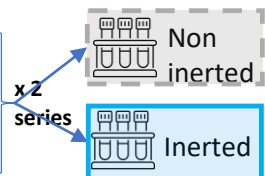
Table 1: Degradation conditions of the Histidine solution (without inerting) prepared according to the formulation of the hospital preparation

### Detection of impurities by reversed phase HPLC-PDA and identification by LC-MS

### Justification for the inerting of Histidine vials

The experimental plan using the stability prediction software based on the Arrhenius law (ASAPPrime®).

- T0: Day 0 (4 samples)
- 45°C: Day 12 (2), Day 15 (2), Day 18 (3), Day 21 (3)
- 55°C: Day 5 (2), Day 8 (2), Day 13 (2), Day 21 (3)
- 65°C: Day 2 (2), Day 4 (2), Day 8 (2), Day 17 (3)
- 75°C: Day 1 (2), Day 2 (2), Day 4 (2), Day 7 (3)
- 90°C: Day 1 (4), Day 2 (5)



- Evaluation of the prospective shelf life for each set.

## RESULTS

### FORCED DEGRADATION

Stress factors	Control	Thermal	Molecular oxidation	Radical oxidation	Photolytic	Alkaline Hydrolysis	Acid Hydrolysis
Stress Conditions	Not Applicable	90 °C	3% H <sub>2</sub> O <sub>2</sub> 60 °C	5% MMPP 60 °C	UV-VIS light 25 °C	1M NaOH 60 °C	1M HCl 60 °C
Aspect	Colorless	Colorless to yellow	Yellow to brown	Yellow to brown + precipitate	Colorless to yellow	Colorless to yellow	Colorless to yellow
Sampling time (Days)	0	2	1	1	21	57	57
Histidine (RT = 5,80 min)	100%	97,96%	94,96%	95,74%	98,69%	97,21%	98,91%
Peak 2 (RRT = 0,62)	0%	0,52%	3,83%	1,27%	0,86%	0,54%	0,43%
Peak 4 (RRT = 2,86)	0%	0,65%	0,47%	2,03%	1,31%	1,34%	0,46%
Sum of peaks	0	15	35	23	31	13	11

Table 2: Examples of results in percentage of normalized area of Histidine DP under various stress factors. (Peak 1 RRT = 0.43, Peak 2 RRT = 0.62, Peak 3 RRT = 2.00, and Peak 4 RRT = 2.80)

### IDENTIFICATION OF A DP THROUGH MS COUPLING

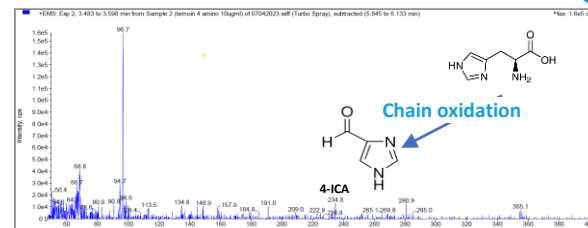


Figure 1: ESI+ MS Scan of 4-Imidazolecarboxaldehyde (4-ICA), specificity of the MRM transition 97/69.

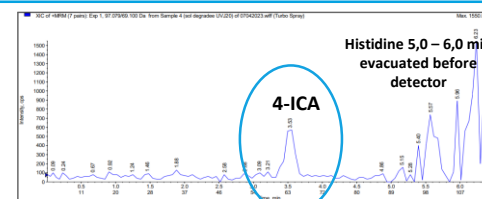


Figure 2: MRM transition 97/69 found at a retention time (RT) of 3.5 minutes in a histidine solution degraded under light for 20 days.

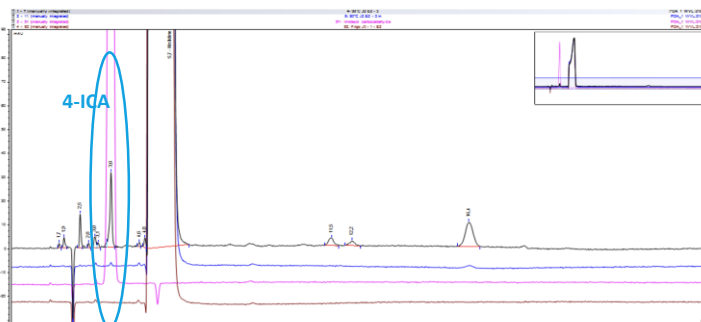


Figure 3: Chromatograms of histidine solutions degraded over 2 days at 90°C, inerted (in blue) and inerted (in black) undegraded histidine solution at Day 0 (in brown), 4-ICA reference (in pink).

## CONCLUSION

The method used in stability studies allows for the visualization and separation of various DPs obtained through forced degradation. The identification of 4-imidazolecarboxaldehyde has provided a representative indicator of histidine solution degradation. The ASAPPrime® program has enabled the prediction of the stability of the histidine formula in less than a month and justified the use of an inerting process for histidine vials.

This study illustrates the analytical methodologies employed in the drug development process.

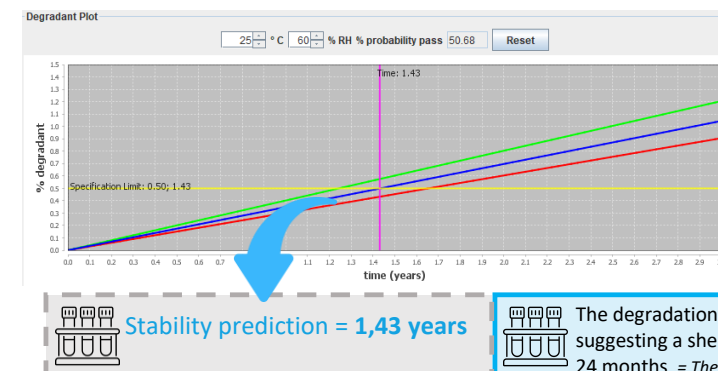
Stress condition	Degradation level
H <sub>2</sub> O <sub>2</sub> Oxidation	+++
MMPP Oxidation	+++
Photolytic	++
Thermal	+
Alkaline	+
Acid	+

- Histidine degradation is observable through changes in the solution's color, (Colorless -> Yellow -> Brown).
- The chromatographic method allows for the separation of a large number of DPs highlighting the method's stability-indicating nature (Figure 3).
- Oxidation is identified as the primary degradation pathway of histidine

The LC-UV analysis revealed that 4-ICA elutes at the same retention time as Peak 2. (Figure 3)  
 The identification of Peak 2 was confirmed through LC-MS coupling. (Figures 1 and 2)

- The peak at 3.5 minutes is indeed 4-ICA. It is planned to replicate the methodology with an oxo-histidine reference, another degradation product mentioned in the literature.<sup>1,2</sup>

### DEGRADATION PREDICTION



- This comparative study demonstrates the effectiveness of inerting in the histidine injectable ampoule manufacturing process

The degradation in the inerted set was very low suggesting a shelf life significantly greater than 24 months = The Histidine/Copper kit shelf life

<sup>1</sup> Mason B, McCracken M, Bures E, Kerwin B. Oxidation of free L-histidine by tert-butylhydroperoxide. Pharm Res. 2010;27(3):447-56  
<sup>2</sup> Méndez-Hurtado J, López R, Suárez D, Menéndez MI. Theoretical study of the oxidation of histidine by singlet oxygen. Chemistry. 2012 Jul 2;18(27):8437-46